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# OLDER LOW-EFFICIENT BOILERS BROWN COAL-FIRED, GENERATED EMISSIONS AND DIFFUSION OF IMISSIONS

At thermooxidizing processes generating energy from natural solid fuels, heat as its main product is accompanied with side products - solid waste and emissions, resulting in unfavourable effects on the environment. This contribution deals with the results of measurements and analyses regarding an old-fashioned type boiler unit fired by solid fuel. It showed low efficiency of older brown coal fired boilers currently used in many small companies. All of them generate pollutants - gaseous as well as solid emissions polluting the environment. Thus, they cause environmental damage in spite of their meeting emission and imission limits within their category. The imission value of tar component at low temperature of flue gases in fireplace and short time of flue gas remaining in fireplaces causes the presence of some components over the permissible imission limit.

Key words: emissions, imissions, brown coal, solid fuel-fired small-size boilers

## 1. Introduction and background

In spite of significant accomplishments that were achieved in the reduction of emissions and imissions in Slovakia during the last decade, there are still many small and midsize sources of emissions that abundantly contribute to the environmental pollution. It is especially old fired heating equipment fuelled with fossil fuels mainly occurring in villages that are not gas connected, but heated by old boilers fuelled with fossil fuels with the heat capacity up to 2 MWth. According to the legislative statutes [1, 2], small sources of pollution do not have determined exact emission limits. That is why entrepreneurs, mainly physical persons, are not obliged to respect certain rules while building such appliances and are not even fined for the air pollution. Midsize sources with lower capacity - up to 2 MWth have only limited concentrations of sulphur and nitrogen oxides, while sulphur oxides are limited by the concentration of 2500 mg.m<sup>-3</sup> only under those circumstances when emission flow of sulphur oxides is above 20 kg an hour, which is not reached by these kinds of appliances. The emission limit of nitrogen oxides in this category of combustion appliances is 650 mg.m<sup>-3</sup>, which is difficult to achieve because with this pollutant, the concentration in the combustion products is quantitatively dependent either on the temperature that is measured in the combustion chamber of the boiler, or on the content of fixed organic nitrogen in the fuel. To create nitrogen oxides from the air by oxidation of the airborne nitrogen are, as a rule, needed temperatures over 1100 °C. However, this temperature cannot be achieved while using small sources fuelled with fossil fuels and smaller boiler units of older types with the heat capacity up to 2 MW. It happens where combustion products from the combustion area usually flow through the flue or directly through the combustion pipe into the chimney. That is why nitrogen oxides in the combustion products whose temperature in older appliances is below 800 °C do not originate from the airborne nitrogen but from fuel only. Nitrogen in fossil natural fuel can be found especially in heterocyclic components of compounds and also in wood as proteinaceous residues and amino derivative polysaccharides. There is about 0.7 % of nitrogen in wood, while the occurrence of nitrogen in coal is more variable, usually a bit higher, however, it depends on the locality of mining. That is why the content of nitrogen oxides in the combustion products of coal is usually higher than in the combustion products of wood. The low temperature of the combustion process as well as a short time of flue gas remaining during the combustion of the fossil fuels in the chamber, cause a defective combustion of the gassed fuel. This will result in poor quality of combustion products which can even be visually seen as the presence of a higher portion of soot in combustion products that escape through the chimney into the air, or particularly when analytically evaulating combustion products components. Gaseous combustion products from organic nitrogen are found as nitrogen oxide if the temperature is low. Oxidation of nitrogen oxide materialises only in the atmosphere.

In particular, it is pointed out that there are inappropriate conditions for inlets of primary and secondary air into firing space with regard to the amount of gasified fuel. Inflow of air and directions of its flow are also important. Poor mixing gases in the firing space causes inappropriate combustion, which will give a high content of carbon monoxides as well atmospheric oxygen in different locations of the firing space. At the older low-efficient boilers there is a characteristic short flow distance of combustion gases in the firing

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space, which results in influencing also its good mixing with air in the firing space. It results in a low thermal efficiency, but also in reduced heating value.

Relatively low temperature of fuel gasification as a short time of gassed products remaining in combustion area cause defective combustion. Resulting in dark smoke during such combustion conditions will cause following reactions: there will be an emission of inorganic residues (in the combustion area) along with condensed aromatic organic compounds, often with the content of fixed oxygen that make these inorganic residues visually black. The chemical structure of these substances is very various and also changeable (it depends on conditions of combustion process and combusting fuel) and they are called tarry substances. Many of them belong amongst persistent organic compounds (POC) that are distinguished for their longevity in the atmosphere and their physiological influence on living creatures, also plants and animals is often considered as a carcinogenic, mutagenical, or teratogenic one.

#### 2. Materials and methods

Some measurements were undertaken to find out combustion products and their diffusion into the atmosphere from an operation of a boiler unit VSB-4 made in ŽD Bohumín in 1972 with a capacity of 0.25 MWt per hour, working under usual conditions. There is a combustion product ventilator installed in this boiler that is normally used for inundating. The ventilator was not operating during the measurement of the combustion products. During the measurement of emissions and also at usual operation which is carried out only during the winter heating, brown coal type  $\rm O_2E$  with a content of ash 12.5 % and sulphur up to 1 % was used.

The measurement was carried out in a walled flue pipe and places of the measurement were determined according to the rules of STN ISO 10 780 [5] and STN ISO 9096 [6]. The flow of the combustion products was calculated from measured physical quantities (dynamic, static and atmospheric pressures, temperature of the combustion products) by Pitot-Prandtl tube and digital micromanometre and density of the combustion products was calculated from the chemical structure of the combustion products (containing oxygen, nitrogen and carbon dioxide, or water vapour). Concentration of combustion gases was calculated from the chemical composition of combustion gases (content of oxygen, nitrogen and carbon dioxin, or water steam). The concentrations of oxygen and carbon oxide were determined by a gas analyser FY IMR 1200 with electrochemical sensors, while carbon dioxide was deducted by analogy. The sulphur oxides were determined by the classical method after sampling into 3 % hydroperoxide according to STN 83 4711 [9]. The sample of fossil pollutants was taken by Ströhlein apparatus STG 4E and determined by the gravimetric method according to [6]. The qualitative characteristics of fossil emissions was done by determination of a soluble portion of fossil elements in carbon disulphide and the consequential analysis were determined by the gaseous chromatography method of dividing on a capillary column of medium polarity. The programmed change of column

temperature was used for separation. The graphic record is presented in Figure 1.

The diffusion of a few volatile organic components (VOC) from a defective combustion of coal was determined according to Szabu programme MODIM 1.20 d [12].

Within Szabu programme MODIM 1.20 d [12], the author applied methodology of the MLVH (1979) for interpolation of a wind rose.

#### 3. Results

The results of concentration and flow of the emissions of basic polluting substances from brown coal combustion in older combustion equipment presented in tables 1-3 are indicated under reference conditions according to the legislative directives [10]. The content of soluble substances from isolated SPM by an organic solvent (CS<sub>2</sub>) was 92.5 %, concluding that an average content of inorganic residues in combustion products was 54 mg.m $^{-3}$ <sub>r</sub>. Supposing that during the combustion process while the low temperatures are present, the content of condensed organic substances emitted into the atmosphere of average concentration was 604 m.g<sup>-3</sup>, which is emission flow of 531 g.h $^{-1}$  from one boiler. Even if volatile organic compounds (VOC) are not the subjects of evaluation of the combustion product quality according to contemporary applicable legislation, it is interesting to notice their quantity as well as quality arising from the process of an inferior combustion action. The highest concentration of vaporous organic substances of the human breathing zone during average climatic conditions (3.1 m.s<sup>-1</sup> wind speed, 10 °C air temperature, NE prevailing air circulation) was 10  $\mu$ g.m<sup>-3</sup> at a distance of 450-600 m from the chimney in SW direction during an actual average meteorological situation. The worst concentration of the VOC during unfavourable climatic conditions (calm, stagnation of air layers) is assumed to happen at a distance of 120-320 m from the chimney. In this case the calculated concentration of the VOC in the combustion products was 174 µg.m<sup>-3</sup>. The coefficient "S" used as a criterion for the calculation of minimal height valued at 10  $\mu$ g.m<sup>-3</sup> for naphthalene and its derivatives, while toluene and xylene valued 200  $\mu$ g.m<sup>-3</sup>. These characteristics can be compared with the value of IHk. Owing to the fact that tars from brown coal mainly contain naphthalene and its derivatives, it can be assumed that there is a contravention of imissive limit.

### 4. Discussion

The entrepreneurs of air polluting sources are by laws related to the atmosphere supposed to use BAT (Best Available Technology) when building new mechanisms, i.e. No. 478/2002 [1]. This concept was developed as a directive concerning the atmosphere and known as BATNECC (best available technologies not entailing cost). In the quoted Act the BAT is defined as the best available technology economically achievable, in section § 6, segment 5.



Sample ID

decht1-1

Sample From

Primary Project

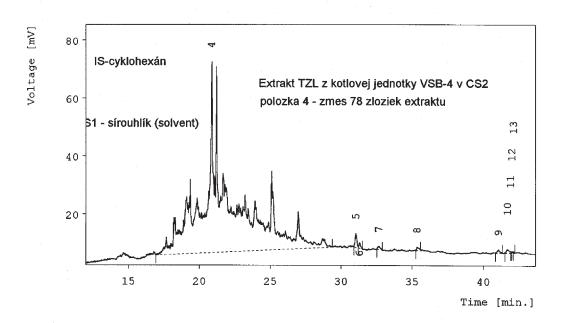
Friday, 21th May, 1999 18:12:04

decht01

Analyst : J.B.

Printed Tuesday, 13th July, 1999 10:27:09

Calibration : (none)



Result Table - Calculation Method Uncal

| Peak | Reten. | Reten. Area |         | W05    | Area   | Height |
|------|--------|-------------|---------|--------|--------|--------|
| No.  | time   | [mV.s]      | [mV]    | [min.] | [%]    | [8]    |
| 2    | 4.147  | 1551.3123   | 679.386 | 0.040  | 17.406 | 89.716 |
| · 4  | 20.887 | 7238.4257   | 65.806  | 0.133  | 81.217 | 8.690  |
| 5    | 31.080 | 42.4467     | 4.740   | 0.147  | 0.476  | 0.626  |
| 6    | 31.333 | 15.9550     | 1.931   | 0.120  | 0.179  | 0.255  |
| 7    | 32.700 | 13.8741     | 1.094   | 0.233  | 0.156  | 0.145  |
| 8    | 35.380 | 11.9369     | 1.061   | 0.220  | 0.134  | 0.140  |
| 9    | 41.113 | 14.3074     | 1.097   | 0.180  | 0.161  | 0.145  |
| 10   | 41.760 | 20.3331     | 1.185   | 0.260  | 0.228  | 0.156  |
| 11   | 41.967 | 0.3798      | 0.157   | 0.040  | 0.004  | 0.021  |
| 12   | 42.073 | 1.5540      | 0.341   | 0.087  | 0.017  | 0.045  |
| 13   | 42.180 | 1.8781      | 0.463   | 0.067  | 0.022  | 0.061  |
| -    | Total  | 8912.4031   | 757.260 |        |        |        |

Fig. 1 The arrangement of VOC which were isolated from solid particles absorbed on filter during SPM sample take-off by the gas chromatography method applied.

Legend: • IS-cyklohexán: cyclohexane • S1 - sírouhlík: carbon disulphide (solvent) • Extrakt TZL z kotlovej jednotky VSB-4 v CS₂: Extract of SPM (Solid Polluting Matters) from VSB-4 boiler unit sample taken from chimney in carbon disulphide (CS2) • Polozka 4 - zmes 78 zloziek extraktu: Item 4 is mixture of 78 components found in an SPM extract.

According to Virčíková [11] the BAT parameters are as follows: emissive limit (or other characteristics if a polluting substance emitted from a source has no emissive limit) + general conditions of operation + technical parameters of operation + precautions to avoid dangerous and accidental

situations + comprehensiveness of air protection solution + security of emissions diffusion + comparability with an air protection standard of developed industrial countries. The BTA criteria are determined by the general obligatory directives dealing with the air protection, technical norms (STN, OTN ŽP, EN, ISO), directives issued by the EU and by the norms of developed countries (e.g. EPA, VDI etc.), and by standard condition of the technology.

The heat production from the fossil fuels, including various kinds of coal, also brings certain obligations for investors in terms



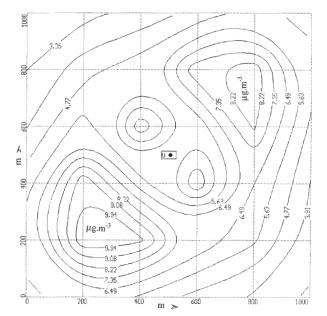


Fig. 2 Isolines of VOC imission concentration gained by brown coal burning technology at low temperature in VSB-4 boiler unit evaluated in the area with average climatic conditions

Fig. 3 Isolines of VOC imission concentration in human breathing zone under inclement climatic conditions for their scattering

Concentration and emission flow, emission limits (by reference conditions – 101.325 kPa, 0  $^{\circ}C$  normalised to 6 % oxygen in waste gas).

Fuel: brown coal

| No. | Polluting mater              | Concentratio | Emission       |                            |
|-----|------------------------------|--------------|----------------|----------------------------|
|     |                              | Measured *)  | emission limit | flow (kg.h <sup>-1</sup> ) |
| 1.  | Solid Polluting Matter (SPM) | 658          | undefined      | 0.579                      |
| 2.  | Sulphur Oxide                | 1406         | 2 5001)        | 1.2373                     |
| 3.  | Nitrogen Oxides              | 268          | 650            | 0.2358                     |
| 4.  | Carbon Oxide                 | 2590         | Undefined      | 2.2792                     |

Explanatory notes: \*) - results are given as arithmetic average of 12 measurements and analysis from two flue-gas off-takes,

Basic statistic characteristics of measured emission source – mass concentration of flue-gas components and volume flow Measurement 1

Table 2

Table 1

| Emission source       | Measured characteristics *1)                              | Arithmetic average (x) | Number of measurements | Maximal value | Minimal value | Decisive deviation (S <sub>x</sub> ) | Error band $L_{1,2}^{*2}$ |
|-----------------------|---|------------------------|------------------------|---------------|---------------|--------------------------------------|---------------------------|
| Boiler K <sub>2</sub> | Flow [m <sup>3</sup> .h <sup>-1</sup> ] (Reference Condit | ons) 882               | 6                      | 940           | 838           | 40                                   | 985 - 780                 |
|                       | Concentration SPM [mg.m                                   | <sup>-3</sup> ] 972    | 6                      | 1063          | 502           | 189                                  | 1340 - 380                |
|                       | Concentration SO <sub>2</sub> [mg.m                       | <sup>-3</sup> ] 1441   | 6                      | 1733          | 1103          | 184                                  | 1913 - 968                |
|                       | Concentration NO <sub>x</sub> [mg.n                       | <sup>-3</sup> ] 238    | 6                      | 256           | 205           | 17                                   | 282 - 194                 |
|                       | Concentration CO [mg.n                                    | $^{-3}$ ] 3195         | 6                      | 3695          | 2420          | 435                                  | 4313 - 2007               |

Explanatory notes: \*1) under reference conditions (dry gas, 0 °C, 101 325 Pa, content of  $O_2 = 6$  % V). Average temperature of flue-gases in chimney was 54 °C

 $<sup>^{1)}</sup>$  - at mass flow  $SO_2$  over 20 kg/h.

<sup>\*2)</sup> error band  $L_{1,2} = x \pm t * S_x$ , where x – arithmetic average, t – coefficient of Student difference,  $S_x$  – decisive deviation



Basic statistic characteristics of measured emission source - mass concentration of flue-gas components and volume flow Measurement 2

Table 3

| Emission source       | Measured characteristics *1)                                   |               | Arithmetic average (x) | Number of measurements | Maximal value | Minimal value | Decisive deviation (S <sub>x</sub> ) | Error band $L_{1,2}^{*2}$ |
|-----------------------|--|---------------|------------------------|------------------------|---------------|---------------|--------------------------------------|---------------------------|
| Boiler K <sub>2</sub> | Flow [m <sup>3</sup> .h <sup>-1</sup> ] (Reference conditions) |               | 877                    | 6                      | 917           | 821           | 30                                   | 954 - 800                 |
|                       | Concentration SPM  | $[mg.m^{-3}]$ | 344                    | 6                      | 410           | 267           | 59                                   | 462 - 226                 |
|                       | Concentration SO <sub>2</sub>                                  | $[mg.m^{-3}]$ | 1371                   | 6                      | 1468          | 1320          | 85                                   | 1590 - 1153               |
|                       | Concentration NO <sub>x</sub>                                  | $[mg.m^{-3}]$ | 299                    | 6                      | 359           | 251           | 34                                   | 386 - 212                 |
|                       | Concentration CO   | $[mg.m^{-3}]$ | 1985                   | 6                      | 2241          | 1827          | 200                                  | 2500 - 1471               |

Explanatory notes: \*1) under reference conditions (dry gas, 0 °C, 101 325 Pa, content of  $O_2 = 6 \% V$ ). Average temperature of flue-gases in chimney was 65 °C

of legislative norms and statutes. The evaluated apparatus for the heat production - VSB-4 boiler unit determined for the combustion of fossil fuel, according to the BAT requirements, fulfils only requirements concerning emissions limits and their diffusion which are not too strict for low efficiency heating equipment. According to the accomplished measurements it is obvious that the entrepreneur did not achieve the required temperature parameters of combustion products in the chimney, resulting in low temperature of the combustion products in the combustion area of boiler and also defective combustion process resulting in high concentration of carbon dioxide and polluting fossil fuels resulting in SPM (Fig. 1). According to the GC analysis, the character of these organic substances is similar to purposely made tars from brown coal, whose main components are aromatic hydrocarbons, heterocyclic oxygenous, nitrogenous, and sulphuric compounds. Their physiological influence on human and other organisms (e.g. animals, plants etc.) differs in comparison with inorganic compounds of ash. According to figures in tables 2 and 3 it is obvious that higher occurrence of the defective combustion products (carbon dioxide, SPM of organic origin) appears during the low temperature combustion.

Therefore, it is very important for every entrepreneur to respect operational directives and requirements to improve operational quality of the combustion processes and reduce the amount of polluting substances in combustion products that are emitted into the atmosphere [13, 14].

#### 5. Conclusion

During brown coal combustion at a low temperature, the emissions of polluting substances are emerged and they contain larger

amounts of defective combustion products, such as carbonic dioxide and organic gases and vapours of higher aromatic hydrocarbons, heterocyclic compounds and their derivatives, which after refrigeration are transformed into soot-like substances that are soluble in organic solvents. Their amount and qualitative composition depends on the type of combusted coal and conditions of combustion process. During the defective combustion process, the emerging VOCs, pollute the atmosphere, increase the amount of smog in the atmosphere, influence production of ground ozone and they can also cause physiological changes and health problems [15–18].

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